

**Official Amendment**

Serial No. – 10/625,886

Docket No. – UVD 0299 IA / UD 268

Amendments to the Claims

The following listing of claims will replace all prior versions, and listings, of claims in the present application:

1. (Currently Amended) A solid corrosion-inhibiting seal for a barrier film formed on a coating selected from anodic coatings, phosphating coatings, or black oxide coatings, the solid corrosion-inhibiting seal comprising a rare earth element and a valence stabilizer combined to form a rare earth/valence stabilizer complex within the solid corrosion-inhibiting seal, wherein the rare earth element is selected from cerium, praseodymium, terbium, or combinations thereof, and at least one rare earth element is in the tetravalent oxidation state.
2. (Original) The corrosion-inhibiting seal of claim 1 wherein the rare earth/valence stabilizer complex has a solubility in water of between about  $5 \times 10^{-1}$  and about  $1 \times 10^{-5}$  moles per liter of cerium, praseodymium, or terbium at about 25°C and about 760 Torr.
3. (Original) The corrosion-inhibiting seal of claim 2 wherein the solubility in water of the rare earth/valence stabilizer complex is between about  $5 \times 10^{-2}$  and about  $5 \times 10^{-5}$  moles per liter of cerium, praseodymium, or terbium at about 25°C and about 760 Torr.
4. (Original) The corrosion-inhibiting seal of claim 1 wherein there is an electrostatic barrier layer around the rare earth/valence stabilizer complex in aqueous solution.
5. (Original) The corrosion-inhibiting seal of claim 1 wherein the rare earth/valence stabilizer complex acts as an ion exchange agent towards corrosive ions.

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6. (Currently Amended) The corrosion-inhibiting seal of claim 1 wherein the barrier film comprises anodic coatings, phosphating coatings, or black oxide coatings comprise a compound selected from oxides, hydroxides, phosphates, carbonates, oxalates, silicates, aluminates, borates, polymers, or combinations thereof.

7. (Original) The corrosion-inhibiting seal of claim 1 wherein the rare earth/valence stabilizer complex has a central cavity containing a cerium, praseodymium, or terbium ion and an additional ion.

8. (Original) The corrosion-inhibiting seal of claim 7 wherein the additional ion is  $B^{+3}$ ,  $Al^{+3}$ ,  $Si^{+4}$ ,  $P^{+5}$ ,  $Ti^{+4}$ ,  $V^{+5}$ ,  $V^{+4}$ ,  $Cr^{+6}$ ,  $Cr^{+3}$ ,  $Mn^{+4}$ ,  $Mn^{+3}$ ,  $Mn^{+2}$ ,  $Fe^{+3}$ ,  $Fe^{+2}$ ,  $Co^{+3}$ ,  $Co^{+2}$ ,  $Ni^{+2}$ ,  $Ni^{+3}$ ,  $Ni^{+4}$ ,  $Cu^{+2}$ ,  $Cu^{+3}$ ,  $Zn^{+2}$ ,  $Ga^{+3}$ ,  $Ge^{+4}$ ,  $As^{+5}$ ,  $As^{+3}$ , or  $Zr^{+4}$ .

9. (Original) The corrosion-inhibiting seal of claim 1 wherein the valence stabilizer is selected from an inorganic valence stabilizer and an organic valence stabilizer.

10. (Original) The corrosion-inhibiting seal of claim 9 wherein the valence stabilizer is the inorganic valence stabilizer selected from molybdates, tungstates, vanadates, niobates, tantalates, tellurates, periodates, iodates, carbonates, antimonates, stannates, phosphates, nitrates, bromates, sulfates, titanates, zirconates, bismuthates, germanates, arsenates, selenates, borates, aluminates, silicates, or combinations thereof.

11. (Original) The corrosion-inhibiting seal of claim 10 wherein the valence stabilizer is the inorganic valence stabilizer selected from molybdates, tungstates, vanadates, niobates, tantalates, tellurates, periodates, iodates, carbonates, antimonates, stannates, phosphates, nitrates, bromates, sulfates, or combinations thereof.

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12. (Withdrawn) The corrosion-inhibiting seal of claim 9 wherein the valence stabilizer is the organic valence stabilizer selected from monoamines; diamines; triamines; tetraamines; pentamines; hexamines; five- or six-membered heterocyclic rings containing one to four nitrogen atoms optionally having additional nitrogen, sulfur, or oxygen binding sites; five- or six-membered heterocyclic rings containing one or two sulfur atoms and having additional nitrogen binding sites; five- or six-membered heterocyclic rings containing one or two oxygen atoms and having additional nitrogen binding sites; (two-, three-, four-, six-, eight-, or ten-)membered nitrogen, nitrogen-sulfur, or nitrogen-oxygen macrocyclics; macrocyclic oligothioketones or dithiolenes; diazenes; thio-, amido-, or imido-derivatives of hypophosphoric, phosphoric, or diphosphoric acids and salts; thio-derivatives of hypophosphoric, phosphoric, or diphosphoric acids and salts; azo compounds, triazenes, formazans, azines, hydrazones, or Schiff Bases containing at least two azo, imine, or azine groups; azo compounds, triazenes, formazans, azines, hydrazones, or Schiff Bases; ortho- (for aryl) or alpha- or beta- (for alkyl) substituted azo compounds, triazenes, formazans, azines, hydrazones, or Schiff Bases; oximes; amidines and imido compounds; dithio ligands; amides; amino acids; N-(thio)acyl 7-aminobenzylidenimines; (thio)hydroxamates; alpha- or ortho-aminothio(di)carboxylic acids and salts; (thio)semicarbazones; (thio)acyl hydrazones; (thio)carbazones; silylaminoalcohols; thioalkyl amines and imines; hydroxyalkyl imines; (thio)aryl amines and imines; guanylureas; guanidinoureas; 2-nitrosophenols; 2-nitrophenols; N-nitrosohydroxylamines; 1,3-monothioketones; monothiomalonamides; 2-thioacylacetamides; 2-acylthioacetamides; dithiodicarbonic diamides; trithiodicarboxylic acids and salts; monothiocarbamates; monothioethers; dithioethers; trithioethers; tetrathioethers; pentathioethers; hexathioethers; disulfides; monophosphines; diphosphines; triphosphines; tetraphosphines; pentaphosphines; hexaphosphines; five- or six-membered heterocyclic rings containing one or two sulfur atoms optionally having additional sulfur, oxygen, or phosphorus binding sites; five- or six-membered heterocyclic rings containing one to three phosphorus atoms optionally having additional phosphorus, nitrogen, oxygen, or sulfur binding sites; five- or six-membered heterocyclic rings

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containing one to four nitrogen atoms and having additional phosphorus binding sites; five- or six-membered heterocyclic rings containing one or two oxygen atoms and optionally having additional sulfur or phosphorus binding sites; (five-, seven-, or nine-)membered nitrogen, nitrogen-sulfur, or nitrogen-oxygen macrocyclics; (two- to ten-)membered sulfur, sulfur-oxygen, or sulfur-phosphorus macrocyclics, not including oligothioketones or dithiolenes; (two- to ten-)membered phosphorus, nitrogen-phosphorus, or oxygen-phosphorus macrocyclics; (two- to ten-)membered oxygen macrocyclics; thio-, amido-, or imido- derivatives of phosphonic and diphosphonic acids and salts containing no sulfur binding sites; amido-, or imido-derivatives of hypophosphoric, phosphoric, or diphosphoric acids and salts containing no sulfur binding sites; dithioperoxydiphosphoramides; dithioperoxydiphosphoric acids and salts; monothioperoxydiphosphoramides; monothioperoxydiphosphoric acids and salts; monothiophosphoric acids; phosphoro(dithioperoxoic) acids and salts; azo compounds, triazenes, formazans, azines, or Schiff Bases; silylamines; silazanes; guanidines and diguanidines; pyridinaldimines; hydrazones; hydramides; nitriles; thioureas and thioamides; ureas and pseudoureas; biurets; monothio ligands; diketone ligands; dithioperoxydicarbonic acids and salts; dithioacyl disulfides; tetrathioperoxydicarbonic diamides; (hexa-, penta-, or tetra-)thioperoxydicarbonic acids and salts; 1,2-dithiolates; rhodanines; dithiocarbimates; (thio)xanthates; S-(alkyl- or aryl-thio)thiocarboxylic acids and salts; phosphinodithioformates; (thio)borates and (thio)boronates; (thio)arsonic acids and salts; (thio)antimonic acids and salts; phosphine and arsine sulfides or oxides; beta-hydroxyketones and -aldehydes; squaric acids and salts; carbonates; carbamates and carbimates; carbazates; imidosulfurous diamides; sulfuriimines; thiocarbonyl and mercapto oximes; 2-nitrothiophenols; 2-nitrido(thio)phenols; acylcyanamides; imidates; 2-amidinoacetates; beta-ketoamines; 3-aminoacrylamides and 3,3-diaminoacrylamides; 3-aminoacrylic acids and salts and 3-hydroxy-3-aminoacrylic acids and salts; 2-nitroanilines; amine and diazine N-oxides; hydrazides and semicarbazides; (amino- or imino-)aryl phosphines; (thio- or hydroxy-)aryl phosphines; arsines; five- or six-membered heterocyclic rings containing one arsenic atom optionally having additional arsenic binding sites;

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(two- to six-)membered arsenic macrocyclics; selenoethers; five- or six-membered heterocyclic rings containing one or two selenium atoms optionally having additional selenium binding sites; (two- to six-)membered selenium macrocyclics; 1,3-diselenoketones; 1,1-diselenolates; diselenocarbamates; selenophosphoric acids and salts; selenocarbonates; cyanide, isocyanide, and cyanamide ligands; nitrosyl and nitrite ligands; azide ligands; thiolates and selenolates; (thio)cyanate ligands; diene or bicyclic or tricyclic hydrocarbon ligands; carbonyl, halogen, oxo, and hydroxo ligands; or combinations thereof.

13. (Withdrawn) The corrosion-inhibiting seal of claim 12 wherein the organic valence stabilizer is selected from monoamines; diamines; triamines; tetraamines; pentamines; hexamines; five- or six-membered heterocyclic rings containing one to four nitrogen atoms optionally having additional nitrogen, sulfur, or oxygen binding sites; five- or six-membered heterocyclic rings containing one or two sulfur atoms and having additional nitrogen binding sites; five- or six-membered heterocyclic rings containing one or two oxygen atoms and having additional nitrogen binding sites; (two-, three-, four-, six-, eight-, or ten-)membered nitrogen, nitrogen-sulfur, or nitrogen-oxygen macrocyclics; macrocyclic oligothioketones or dithiolenes; diazenes; thio-, amido-, or imido-derivatives of hypophosphoric, phosphoric, or diphosphoric acids and salts; azo compounds, triazenes, formazans, azines, hydrazones, or Schiff Bases containing at least two azo, imine, or azine groups; azo compounds, triazenes, formazans, azines, hydrazones, or Schiff Bases; ortho- (for aryl) or alpha- or beta- (for alkyl) substituted azo compounds, triazenes, formazans, azines, hydrazones, or Schiff Bases; oximes; amidines and imido compounds; dithio ligands; amides; amino acids; N-(thio)acyl 7-aminobenzylidenimines; (thio)hydroxamates; alpha- or ortho-aminothio(di)carboxylic acids and salts; (thio)semicarbazones; (thio)acyl hydrazones; (thio)carbazones; silylaminoalcohols; thioalkyl amines and imines; hydroxyalkyl imines; (thio)aryl amines and imines; guanylureas; guanidinoureas; biurets; diketone ligands; amido- or imido-derivatives of hypophosphoric, phosphoric, or diphosphoric acids and salts; beta-hydroxyketones and -aldehydes; squaric acids

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and salts; carbonates; carbamates and carbimates; 2-nitrosophenols; 2-nitrophenols; N-nitrosohydroxylamines; 1,3-monothioketones; monothiomalonamides; 2-thioacetylacetamides; 2-acylthioacetamides; dithiodicarbonic diamides; trithiodicarboxylic acids and salts; monothiocarbamates; (thio)cyanate ligands; or combinations thereof.

14. (Withdrawn) The corrosion-inhibiting seal of claim 12 wherein the organic valence stabilizer is the diazene selected from diazeneformimidamides, diazeneformamides, diazeneformothioamides, diazeneacetimidamides, diazeneacetothioamides, diazeneformimidic acids and salts, diazeneacetimidic acids and salts, diazenecarbothioic acids and salts, diazenecarbodithioic acids and salts, diazeneformimidothioic acids and salts, diazenecarboxaldehydes, diazeneformothioaldehydes, diazeneacetaldehydes, diazeneacetothioaldehydes, diazenediformamides, diazenediformothioamides, diazenediacetamides, diazenediacetothioamides, diazeneacetimidothioic acids and salts, imido diazenes, diazenediformimidamides, diazenediacetimidamides, diazenediformimidic acids and salts, diazenediacetimidic acids and salts, diazenediformimidothioic acids and salts, diazenediacetothioic acids and salts, diazenediacarbothioic acids, diazenediacarbodithioic acids, diazeneformic acids, diazenediformic acids, diazeneacetic acids, diazenediacetic acids, diazenediformaldehydes, diazenediformothioaldehydes, diazenediacetaldehydes, diazenediacetothioaldehydes, diimido diazenes, or combinations thereof.

15. (Withdrawn) The corrosion-inhibiting seal of claim 12 wherein the organic valence stabilizer is the thio-, amido-, or imido-derivative of hypophosphoric, phosphoric, or diphosphoric acids and salts selected from phosphoramidimidic triamides, phosphoramidimidic acids and salts, phosphorodiamidimidic acids and salts, phosphorodiamidimidothioic acids and salts, phosphoramidimidothioic acids and salts, phosphorodiamidimidodithioic acids and salts, phosphoramidimidodithioic acids and salts, (di- or mono-)thiohypophosphoric acids and salts, (di- or mono-)thiohypophosphoramides, phosphoramidic acids and salts, phosphorimidic acids

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and salts, (di- or mono-)thioimidodiphosphoric acids and salts, (di- or mono-)thiohydrazidodiphosphoric acids and salts, (di- or mono-)thioimidodiphosphoramides, (di- or mono-)thiohydrazidodiphosphoramides, phosphoric triamides, (di- or mono-)thiodiphosphoramides, (di- or mono-)thiodiphosphoric acids and salts, (tetra-, tri-, di-)thiophosphoric acids and salts, phosphoro(dithioperoxo)(mono-, di-, or tri-)thioic acids and salts, phosphorimido(mono-, di-, or tri-)thioic acids and salts, phosphorothioic triamides, phosphoramido(mono, di- or tri-)thioic acids and salts, phosphorodiamido(mono, di- or tri-)thioic acids and salts, or combinations thereof.

16. (Withdrawn) The corrosion-inhibiting seal of claim 12 wherein the organic valence stabilizer is the ortho- (for aryl) or alpha- or beta- (for alkyl) substituted azo compound, triazene, formazan, azine, hydrazone, or Schiff Base substituent selected from amino, imino, oximo, diazeno, hydrazido, thiol, mercapto, thiocarbonyl, hydroxy, carbox, and carbonyl substituents, or combinations thereof.

17. (Withdrawn) The corrosion-inhibiting seal of claim 12 wherein the organic valence stabilizer is the oxime selected from monooximes, dioximes, carbonyl oximes, imine oximes, hydroxy oximes, amino oximes, amido oximes, hydrazone oximes, azo oximes, or combinations thereof.

18. (Withdrawn) The corrosion-inhibiting seal of claim 12 wherein the organic valence stabilizer is the amidine and imido compound selected from amidines, diamidines, biguanides, biguanidines, diamidinomethanes, imidoguanidines, amidinoguanidines, diformamidine oxides, diformamidine sulfides, diformamidine disulfides, imidodicarbonimidic acids and salts, diimidodicarbonimidic acids and salts, thioimidodicarbonimidic acids and salts, thiodiimidodicarbonimidic acids and salts, diimidoylimines, diimidoylhydrazides,

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imidosulfamides, diimidosulfamides, O-amidinocarbamates, O- or S-amidino(mono-, di-, or peroxy-)thiocarbamates, N-hydroxy(or N,N'-dihydroxy)amidines, diimidosulfuric acids and salts, or combinations thereof.

19. (Withdrawn) The corrosion-inhibiting seal of claim 14 wherein the organic valence stabilizer is the dithio ligand selected from dithioimidodialdehydes, dithiohydrazidodialdehydes, dithioimidodicarbonic acids and salts, dithiohydrazidodicarbonic acids and salts, 1,3-dithioketones, 1,2-dithioketones, dithiomalonamides, 2-thioacylthioacetamides, dithioacyl sulfides, trithiodicarbonic diamides, (penta-, tetra-, tri-)thiodicarbonic acids and salts, beta-mercaptopthioketones and -aldehydes, N-(aminomethylthiol)thioureas, dithioxamides, 1,1-dithiolates, (di- or per-)thiomonocarboxylic acids and salts, (tetra- or per-)thiodicarboxylic acids and salts, (di-, tri-, or per-)thiocarbonates, dithiocarbamates (including N-hydroxydithiocarbamates and N-mercaptopdithiocarbamates), dithiocarbazates; or combinations thereof.

20. (Withdrawn) The corrosion-inhibiting seal of claim 14 wherein the organic valence stabilizer is the amide selected from monoamides, lactams, amidinoamides, guanidinoamides, imidoylamides, polyamides, polylactams, or combinations thereof.

21. (Withdrawn) The corrosion-inhibiting seal of claim 14 wherein the organic valence stabilizer is the thio-, amido-, or imido-derivative of phosphonic and diphosphonic acids and salts selected from phosphonitrile amides, phosphonimidic diamides, phosphonamidimidic acids and salts, phosphonamidimidothioic acids and salts, dithioimidodiphosphonic acids and salts, dithiohydrazidodiphosphonic acids and salts, dithioimidodiphosphonamides, dithiohydrazidodiphosphonamides, dithiodiphosphonamides, dithiodiphosphonic acids and salts, dithioperoxydiphosphonamides, dithioperoxydiphosphonic acids and salts, (di- and tri-

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)thiophosphonic acids and salts, phosphono(dithioperoxo)thioic acids and salts, phosphono(dithioperoxo)dithioic acids and salts, phosphonimidothioic acids and salts, phosphonimidodithioic acids and salts, phosphonothioic acids and salts, phosphonanidothioic acids and salts, phosphonamidimidodithioic acids and salts, monothioimidodiphosphonic acids and salts, monothiohydrazidodiphosphonic acids and salts, monothioimidodiphosphonamides, monothiohydrazidodiphosphonamides, monothiodiphosphonamides, monothiodiphosphonic acids and salts, monothioperoxydiphosphonamides, monothioperoxydiphosphonic acids and salts, monothiophosphonic acids and salts, phosphono(dithioperoxoic) acids and salts, or combinations thereof.

22. (Withdrawn) The corrosion-inhibiting seal of claim 12 wherein the organic valence stabilizer is the amido-, or imido-derivative of hypophosphoric, phosphoric, or diphosphoric acids and salts containing no sulfur binding sites selected from hypophosphoric acids and salts, hypophosphoramides, imidodiphosphoric acids and salts, hydrazidodiphosphoric acids and salts, imidodiphosphoramides, hydrazidodiphosphoramides, diphosphoramides, or combinations thereof.

23. (Withdrawn) The corrosion-inhibiting seal of claim 12 wherein the organic valence stabilizer is the amido-, or imido-derivative of phosphonic or diphosphonic acids and salts containing no sulfur binding sites selected from imidodiphosphonic acids and salts, hydrazidodiphosphonic acids and salts, imidodiphosphonamides, hydrazidodiphosphonamides, diphosphonamides, phosphonimidic acids and salts, phosphonamidic acids and salts, phosphonic diamides, or combinations thereof.

24. (Withdrawn) The corrosion-inhibiting seal of claim 12 wherein the organic valence stabilizer is the thiourea and thioamide selected from thioureas, thiocarboxamides, thioacylthioureas, acylthioureas, thioacylureas, thioaroylthioureas, aroylthioureas,

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thioaroylureas, thioimidates, thioguanylureas, guanidinothioureas, amidinothioamides, guanidinothioamides, imidoylthioamides, 3-aminothioacrylamides, thiohydrazides, thiosemicarbazides, (mono- and di-)thiobiurets, (mono- and di-)thioisobiurets, (mono- and di-)thiobiureas, N-(aminomethylol)thioureas, N-(aminomethylthiol)ureas, beta-mercaptopcarboxamides, or combinations thereof.

25. (Withdrawn) The corrosion-inhibiting seal of claim 12 wherein the organic valence stabilizer is the biuret selected from biurets, triurets, isobiurets, biureas, triureas, acylureas, aroylureas, N-(aminomethylol)ureas, or combinations thereof.

26. (Withdrawn) The corrosion-inhibiting seal of claim 12 wherein the organic valence stabilizer is the monothio ligand selected from beta-aminothiones, 3-aminothioacrylic acids and salts, 3-mercaptop-3-aminothioacrylic acids and salts, N-thioacyl benzylidenimines, thioimidodialdehydes, thiohydrazidodialdehydes, thioimidodicarbonic acids and salts, thiohydridodicarbonic acids and salts, 1,2-monothioketones, trithioperoxydicarbonic diamides, dithioperoxydicarbonic diamides, dithiodicarbonic acids and salts, trithioperoxydicarbonic acids and salts, beta-hydroxythioketones, beta-hydroxythioaldehydes, beta-mercaptopketones, beta-mercptoaldehydes, monothiooxamides, beta-mercaptopcarboxylic acids and salts, beta-mercaptopthiocarboxylic acids and salts, beta-hydroxythiocarboxylic acids and salts, S-alkylthiocarboxylic acids and salts, S-arylthiocarboxylic acids and salts, S-alkyldisulfidocarboxylic acids and salts, S-aryldisulfidocarboxylic acids and salts, monothiomonocarboxylic acids and salts, dithiodicarboxylic acids and salts, monothiocarbonates, monothiocarbazates, monothiocarbimates, mercaptoalcohols, silylmercaptoalcohols, or combinations thereof.

27. (Withdrawn) The corrosion-inhibiting seal of claim 12 wherein the organic valence stabilizer is the diketone ligand selected from imidodialdehydes, hydrazidodialdehydes,

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imidodicarbonic acids and salts, hydrazidodicarbonic acids and salts, imidodisulfamic acids and salts, imidodisulfuric acids and salts, 1,3-diketones, 1,2-diketones, malonamides, 2-acylacetamides, monothiodicarbonic diamides, monothiodicarbonic acids and salts, dithioperoxydicarbonic acids and salts, trithionic acids and salts, oxamides, dicarboxylic acids, or combinations thereof.

28. (Withdrawn) The corrosion-inhibiting seal of claim 12 wherein the organic valence stabilizer is the S-(alkyl- or aryl-thio)thiocarboxylic acid and salt selected from S-(alkylthio)thiocarboxylic acids and salts, S-(arylthio)thiocarboxylic acids and salts, S,S-thiobisthiocarboxylic acids and salts, S-(alkyldisulfido)thiocarboxylic acids and salts, S-(aryldisulfido)thiocarboxylic acids and salts, S,S'-disulfidobisthiocarboxylic acids and salts, or combinations thereof.

29. (Withdrawn) The corrosion-inhibiting seal of claim 12 wherein the organic valence stabilizer is the phosphine and arsine sulfide or oxide selected from phosphine P-sulfides, aminophosphine sulfides, arsine As-sulfides, aminoarsine sulfides, phosphine P-oxides, aminophosphine oxides, arsine As-oxides, aminoarsine oxides, or combinations thereof.

30. (Withdrawn) The corrosion-inhibiting seal of claim 12 wherein the solubility in water of the rare earth/valence stabilizer complex is adjusted by an addition of a substituent group on the organic valence stabilizer.

31. (Withdrawn) The corrosion-inhibiting seal of claim 30 wherein the solubility in water is increased by the addition of the substituent group selected from sulfonate groups (-SO<sub>3</sub><sup>-</sup>), carboxyl groups (-CO<sub>2</sub>-), hydroxyl groups (-OH), ester groups (-CO<sub>2</sub>-), carbonyl groups (=C=O), amine groups (-NH<sub>2</sub>), nitrosamine groups (=N-N=O), carbonylnitrene groups (-CO-N), sulfoxide

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groups ( $=\text{S}=\text{O}$ ), sulfone groups ( $=\text{S}[\text{=O}]_2$ ), sulfinyl groups ( $-\text{N}=\text{S}=\text{O}$ ), sulfodiimines ( $=\text{S}[\text{=NH}]_2$ ), sulfonyl halide groups ( $-\text{S}[\text{=O}]_2\text{X}$ ), sulfonamide groups ( $-\text{S}[\text{=O}]_2\text{NH}_2$ ), monohalosulfonamide groups ( $-\text{S}[\text{=O}]_2\text{NH}\text{X}$ ), dihalosulfonamide groups ( $-\text{S}[\text{=O}]_2\text{MX}_2$ ), halosulfonate groups ( $-\text{S}[\text{=O}]_2\text{OX}$ ), halosulfonate amide groups ( $=\text{N}-\text{S}[\text{=O}]_2\text{X}$ ), aminosulfonate groups ( $=\text{N}-\text{S}[\text{=O}]_2\text{OH}$ ), iminosulfonate groups ( $-\text{N}[\text{SO}_3^-]_{1-2}$ ), phosphonate groups ( $-\text{PO}_3^{2-}$ ), phosphonamide groups ( $-\text{PO}_2\text{NH}_2^-$ ), phosphondiamide groups ( $-\text{PO}[\text{NH}_2]_2$ ), aminophosphonate groups ( $=\text{N}-\text{PO}_3^{2-}$ ), iminophosphonate groups ( $-\text{N}[\text{PO}_3^{2-}]_{1-2}$ ), or combinations thereof.

32. (Withdrawn) The corrosion-inhibiting seal of claim 30 wherein the solubility in water is decreased by the addition of the substituent group selected from nitro groups ( $-\text{NO}_2$ ),

perfluoroalkyl groups ( $-\text{C}_x\text{F}_{2x+1}$ ), perchloroalkyl groups ( $-\text{C}_x\text{Cl}_{2x+1}$ ), nitramine groups ( $=\text{N}-\text{NO}_2$ ), thioketone groups ( $=\text{C}=\text{S}$ ), sulfenyl halide groups ( $-\text{S}-\text{X}$ ), sulfur dihaloimide groups ( $-\text{N}=\text{SX}_2$ ), or combinations thereof.

33. (Withdrawn) The corrosion-inhibiting seal of claim 12 wherein an electrostatic barrier layer of the rare earth/valence stabilizer complex is adjusted by an addition of a substituent group on the organic valence stabilizer.

34. (Withdrawn) The corrosion-inhibiting seal of claim 33 wherein the electrostatic barrier layer is increased by the addition of the substituent group selected from ketones ( $=\text{C}=\text{O}$ ), thioketones ( $=\text{C}=\text{S}$ ), amides ( $-\text{C}[\text{=O}]-\text{NR}_2$ ), thioamides ( $-\text{C}[\text{=S}]-\text{NR}_2$ ), nitriles or cyano groups ( $-\text{CN}$ ), isocyanides ( $-\text{NC}$ ), nitroso groups ( $-\text{N}=\text{O}$ ), thionitroso groups ( $-\text{N}=\text{S}$ ), nitro groups ( $-\text{NO}_2$ ), azido groups ( $-\text{N}_3$ ), cyanamide or cyanonitrene groups ( $=\text{N}-\text{CN}$ ), cyanate groups ( $-\text{O}-\text{CN}$ ), isocyanate groups ( $-\text{N}=\text{C}=\text{O}$ ), thiocyanate groups ( $-\text{S}-\text{CN}$ ), isothiocyanate groups ( $-\text{N}=\text{C}=\text{S}$ ), nitrosamine groups ( $-\text{N}-\text{N}=\text{O}$ ), thionitrosamine groups ( $=\text{N}-\text{N}=\text{S}$ ), nitramine groups ( $=\text{N}-\text{NO}_2$ ), thionitramine groups ( $=\text{N}-\text{NS}_2$ ), carbonylnitrene groups ( $-\text{CO}-\text{N}$ ), thiocarbonylnitrene groups ( $-\text{C(=S)}-\text{N}$ ), or combinations thereof.

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CS-N), sulfenyl halides (-S-X), sulfoxides (=S=O), sulfones (=S[=O]<sub>2</sub>), sulfinyl groups (-N=S=O), thiosulfinyl groups (-N=S=S), sulfenyl thiocyanato groups (-S-S-CN), sulfenyl cyanato groups (-S-O-CN), sulfodiimine groups (=S[=NH]<sub>2</sub>), sulfur dihaloimido groups (-N=SX<sub>2</sub>), sulfur oxide dihaloimido groups (-N=S[=O]X<sub>2</sub>), aminosulfur oxide trihalide groups (=N-S[=O]X<sub>3</sub>), sulfonyl azide groups (-S[=O]<sub>2</sub>N<sub>3</sub>), sulfonyl thiocyanate groups (-S[=O]<sub>2</sub>SCN), sulfonyl cyanate groups (-S[=O]<sub>2</sub>OCN), sulfonyl cyanide groups (-S[=O]<sub>2</sub>CN), halosulfonate groups (-S[=O]<sub>2</sub>OX), phosphonyl thiocyanate groups (-P[=O]OHSCN), phosphonyl cyanate groups (-P[=O]OHOCN), phosphonyl cyanide groups (-P[=O]OHCN), or combinations thereof.

35. (Original) The corrosion-inhibiting seal of claim 1 further comprising a solubility control agent.

36. (Original) The corrosion-inhibiting seal of claim 35 wherein the solubility control agent is a cationic solubility control agent or an anionic solubility control agent.

37. (Original) The corrosion-inhibiting seal of claim 36 wherein the solubility control agent is the cationic solubility control agent selected from H<sup>+</sup>; Li<sup>+</sup>; Na<sup>+</sup>; K<sup>+</sup>; Rb<sup>+</sup>; Cs<sup>+</sup>; NH<sub>4</sub><sup>+</sup>; Mg<sup>+2</sup>; Ca<sup>+2</sup>; Sr<sup>+2</sup>; Be<sup>+2</sup>; Ba<sup>+2</sup>; Y<sup>+3</sup>; La<sup>+3</sup>; Ce<sup>+3</sup>; Ce<sup>+4</sup>; Nd<sup>+3</sup>; Pr<sup>+3</sup>; Sc<sup>+3</sup>; Sm<sup>+3</sup>; Eu<sup>+3</sup>; Eu<sup>+2</sup>; Gd<sup>+3</sup>; Tb<sup>+3</sup>; Dy<sup>+3</sup>; Ho<sup>+3</sup>; Er<sup>+3</sup>; Tm<sup>+3</sup>; Yb<sup>+3</sup>; Lu<sup>+3</sup>; Ti<sup>+4</sup>; Zr<sup>+4</sup>; Ti<sup>+3</sup>; Hf<sup>+4</sup>; Nb<sup>+5</sup>; Ta<sup>+5</sup>; Nb<sup>+4</sup>; Ta<sup>+4</sup>; V<sup>+5</sup>; V<sup>+4</sup>; V<sup>+3</sup>; Mo<sup>+6</sup>; W<sup>+6</sup>; Mo<sup>+5</sup>; W<sup>+5</sup>; Mo<sup>+4</sup>; W<sup>+4</sup>; Cr<sup>+3</sup>; Mn<sup>+2</sup>; Mn<sup>+3</sup>; Mn<sup>+4</sup>; Fe<sup>+2</sup>; Fe<sup>+3</sup>; Co<sup>+2</sup>; Co<sup>+3</sup>; Ni<sup>+2</sup>; Ni<sup>+3</sup>; Ni<sup>+4</sup>; Ru<sup>+2</sup>; Ru<sup>+3</sup>; Ru<sup>+4</sup>; Rh<sup>+3</sup>; Ir<sup>+3</sup>; Rh<sup>+2</sup>; Ir<sup>+2</sup>; Pd<sup>+4</sup>; Pt<sup>+4</sup>; Pd<sup>+2</sup>; Pt<sup>+2</sup>; Os<sup>+4</sup>; Cu<sup>+</sup>; Cu<sup>+2</sup>; Cu<sup>+3</sup>; Ag<sup>+</sup>; Ag<sup>+2</sup>; Ag<sup>+3</sup>; Au<sup>+</sup>; Au<sup>+2</sup>; Au<sup>+3</sup>; Zn<sup>+2</sup>; Cd<sup>+2</sup>; Hg<sup>+</sup>; Hg<sup>+2</sup>; Al<sup>+3</sup>; Ga<sup>+3</sup>; Ga<sup>+</sup>; In<sup>+3</sup>; In<sup>+</sup>; Tl<sup>+3</sup>; Tl<sup>+</sup>; Ge<sup>+2</sup>; Sn<sup>+4</sup>; Sn<sup>+2</sup>; Pb<sup>+4</sup>; Pb<sup>+2</sup>; Sb<sup>+3</sup>; Sb<sup>+5</sup>; As<sup>+3</sup>; As<sup>+5</sup>; Bi<sup>+3</sup>; Bi<sup>+5</sup>; organic compounds containing at least one N<sup>+</sup> site; organic compounds containing at least one phosphonium site; organic compounds containing at least one arsonium site; organic compounds containing at least one stibonium site; organic compounds containing at least one oxonium site; organic compounds

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containing at least one sulfonium site; organic compounds containing at least one selenonium site; organic compounds containing at least one iodonium site; quaternary ammonium compounds having a formula  $NR_4^+$ , where R is an alkyl, aromatic, or acyclic organic constituent; or combinations thereof.

38. (Original) The corrosion-inhibiting seal of claim 37 wherein the cationic solubility control agent is selected from  $H^+$ ;  $Li^+$ ;  $Na^+$ ;  $K^+$ ;  $Rb^+$ ;  $Cs^+$ ;  $NH_4^+$ ;  $Mg^{+2}$ ;  $Ca^{+2}$ ;  $Sr^{+2}$ ;  $Y^{+3}$ ;  $La^{+3}$ ;  $Ce^{+3}$ ;  $Ce^{+4}$ ;  $Nd^{+3}$ ;  $Pr^{+3}$ ;  $Sc^{+3}$ ;  $Sm^{+3}$ ;  $Eu^{+3}$ ;  $Eu^{+2}$ ;  $Gd^{+3}$ ;  $Tb^{+3}$ ;  $Dy^{+3}$ ;  $Ho^{+3}$ ;  $Er^{+3}$ ;  $Tm^{+3}$ ;  $Yb^{+3}$ ;  $Lu^{+3}$ ;  $Ti^{+4}$ ;  $Zr^{+4}$ ;  $Ti^{+3}$ ;  $Hf^{+4}$ ;  $Nb^{+5}$ ;  $Ta^{+5}$ ;  $Nb^{+4}$ ;  $Ta^{+4}$ ;  $Mo^{+6}$ ;  $W^{+6}$ ;  $Mo^{+5}$ ;  $W^{+5}$ ;  $Mo^{+4}$ ;  $W^{+4}$ ;  $Mn^{+2}$ ;  $Mn^{+3}$ ;  $Mn^{+4}$ ;  $Fe^{+2}$ ;  $Fe^{+3}$ ;  $Co^{+2}$ ;  $Co^{+3}$ ;  $Ru^{+2}$ ;  $Ru^{+3}$ ;  $Ru^{+4}$ ;  $Rh^{+3}$ ;  $Ir^{+3}$ ;  $Rh^{+2}$ ;  $Ir^{+2}$ ;  $Pd^{+4}$ ;  $Pt^{+4}$ ;  $Pd^{+2}$ ;  $Pt^{+2}$ ;  $Cu^+$ ;  $Cu^{+2}$ ;  $Cu^{+3}$ ;  $Ag^+$ ;  $Ag^{+2}$ ;  $Ag^{+3}$ ;  $Au^+$ ;  $Au^{+2}$ ;  $Au^{+3}$ ;  $Zn^{+2}$ ;  $Al^{+3}$ ;  $Ga^{+3}$ ;  $Ga^+$ ;  $In^{+3}$ ;  $In^+$ ;  $Ge^{+4}$ ;  $Ge^{+2}$ ;  $Sn^{+4}$ ;  $Sn^{+2}$ ;  $Sb^{+3}$ ;  $Sb^{+5}$ ;  $Bi^{+3}$ ;  $Bi^{+5}$ ; organic compounds containing at least one  $N^+$  site; organic compounds containing at least one phosphonium site; organic compounds containing at least one stibonium site; organic compounds containing at least one oxonium site; organic compounds containing at least one sulfonium site; organic compounds containing at least one iodonium site; quaternary ammonium compounds having a formula  $NR_4^+$ , where R is an alkyl, aromatic, or acyclic organic constituent; or combinations thereof.

39. (Withdrawn) The corrosion-inhibiting seal of claim 36 wherein the solubility control agent is the anionic solubility control agent selected from fluorotitanates, chlorotitanates, fluorozirconates, chlorozirconates, fluoroniobates, chloroniobates, fluorotantalates, chlorotantalates, molybdates, tungstates, permanganates, fluoromanganates, chloromanganates, fluoroferrates, chloroferrates, fluorocobaltates, chlorocobaltates, fluorozincates, chlorozincates, borates, fluoroborates, fluoroaluminates, chloroaluminates, carbonates, silicates, fluorosilicates, fluorostannates, nitrates, nitrites, azides, cyanamides, phosphates, phosphites, phosphonates, phosphinates, thiophosphates, thiophosphites, thiophosphonates, thiophosphinates, fluorophosphates, fluoroantimonates, chloroantimonates, sulfates, sulfites, sulfonates,

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thiosulfates, dithionites, dithionates, fluorosulfates, tellurates, fluorides, chlorides, chlorates, perchlorates, bromides, bromates, iodides, iodates, periodates, heteropolyanions, ferricyanides, ferrocyanides, cyanocobaltates, cyanocuprates, cyanomanganates, cyanates, cyanatoferates, cyanatocobaltates, cyanatocuprates, cyanatomanganates, thiocyanates, thiocyanatoferates, thiocyanatocobaltates, thiocyanatocuprates, thiocyanatomanganates, cyanamides, cyanamidoferates, cyanamidocobaltates, cyanamidocuprates, cyanamidomanganates, nitritoferates, nitritocobaltates, azides, (thio)carboxylates, di(thio)carboxylates, tri(thio)carboxylates, tetra(thio)carboxylates, (thio)phenolates, di(thio)phenolates, tri(thio)phenolates, tetra(thio)phenolates, (thio)phosphonates, di(thio)phosphonates, tri(thio)phosphonates, (thio)phosphonamides, di(thio)phosphonamides, tri(thio)phosphonamides, amino(thio)phosphonates, diamino(thio)phosphonates, triamino(thio)phosphonates, imino(thio)phosphonates, diimino(thio)phosphonates, (thio)sulfonates, di(thio)sulfonates, tri(thio)sulfonates, (thio)sulfonamides, di(thio)sulfonamides, tri(thio)sulfonamides, amino(thio)sulfonates, diamino(thio)sulfonates, triamino(thio)sulfonates, imino(thio)sulfonates, diimino(thio)sulfonates, (thio)borates, di(thio)borates, (thio)boronates, organic silicates, stibonates, cyanides, cyanochromates, cyanonickelates, cyanatochromates, cyanatonickelates, thiocyanatochromates, thiocyanatonickelates, cyanamidochromates, cyanamidonickelates, nitritonickelates, arsonates, diarsonates, triarsonates, organic selenates, diselenates, triselenates, arsenates, arsenites, fluoroarsenates, chloroarsenates, selenates, selenites, fluorothallates, chlorothallates, iodomercury anions, chloromercurates, bromomercurates, osmates, fluoronickelates, chromates, Reinecke's salt, vanadates, or combinations thereof.

40. (Withdrawn) The corrosion-inhibiting seal of claim 39 wherein the anionic solubility control agent is selected from fluorotitanates, chlorotitanates, fluorozirconates, chlorozirconates, fluoroniobates, chloroniobates, fluorotantalates, chlorotantalates, molybdates, tungstates, permanganates, fluoromanganates, chloromanganates, fluoroferates, chloroferrates, fluorocobaltates, chlorocobaltates, fluorozincates, chlorozincates, borates, fluoroborates,

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fluoroaluminates, chloroaluminates, carbonates, silicates, fluorosilicates, fluorostannates, nitrates, nitrites, azides, cyanamides, phosphates, phosphites, phosphonates, phosphinites, thiophosphates, thiophosphites, thiophosphonates, thiophosphinites, fluorophosphates, fluoroantimonates, chloroantimonates, sulfates, sulfites, sulfonates, thiosulfates, dithionites, dithionates, fluorosulfates, tellurates, fluorides, chlorides, chlorates, perchlorates, bromides, bromates, iodides, iodates, periodates, heteropolyanions, ferricyanides, ferrocyanides, cyanocobaltates, cyanocuprates, cyanomanganates, cyanates, cyanatoferates, cyanatocobaltates, cyanatocuprates, cyanatomanganates, thiocyanates, thiocyanatoferates, thiocyanatocobaltates, thiocyanatocuprates, thiocyanatomanganates, cyanamides, cyanamidoferates, cyanamidocobaltates, cyanamidocuprates, cyanamidomanganates, nitritoferrates, nitritocobaltates, azides, (thio)carboxylates, di(thio)carboxylates, tri(thio)carboxylates, tetra(thio)carboxylates, (thio)phenolates, di(thio)phenolates, tri(thio)phenolates, tetra(thio)phenolates, (thio)phosphonates, di(thio)phosphonates, tri(thio)phosphonates, (thio)phosphonamides, di(thio)phosphonamides, tri(thio)phosphonamides, amino(thio)phosphonates, diamino(thio)phosphonates, triamino(thio)phosphonates, imino(thio)phosphonates, diimino(thio)phosphonates, (thio)sulfonates, di(thio)sulfonates, tri(thio)sulfonates, (thio)sulfonamides, di(thio)sulfonamides, tri(thio)sulfonamides, amino(thio)sulfonates, diamino(thio)sulfonates, triamino(thio)sulfonates, imino(thio)sulfonates, diimino(thio)sulfonates, (thio)borates, di(thio)borates, (thio)boronates, organic silicates, stibonates, or combinations thereof.

41. (Original) The corrosion-inhibiting seal of claim 1 further comprising a lubricity agent.

42. (Original) The corrosion-inhibiting seal of claim 41 wherein the lubricity agent is selected from molybdenum disulfide, fluorinated hydrocarbons, perfluorinated hydrocarbons, graphite, soft metals, polymers, or combinations thereof.

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43. (Original) The corrosion-inhibiting seal of claim 42 wherein the lubricity agent is the soft metal selected from tin, indium, silver, or combinations thereof.

44. (Original) The corrosion-inhibiting seal of claim 1 wherein the corrosion-inhibiting seal is colored.

45. (Original) The corrosion-inhibiting seal of claim 44 further comprising an agent which improves color-fastness of the corrosion-inhibiting seal.

46. (Original) The corrosion-inhibiting seal of claim 45 wherein the agent which improves color-fastness is selected from an active UV blocker, a passive UV blocker, a brightener, or a combination thereof.

47. (Original) The corrosion-inhibiting seal of claim 46 wherein the agent which improves color-fastness is the active UV blocker selected from carbon black, graphite, phthalocyanines, or combinations thereof.

48. (Withdrawn) The corrosion-inhibiting seal of claim 46 wherein the agent which improves color-fastness is the passive UV blocker selected from titanium oxide, tin oxide, lead oxide, silicon oxide, silicates, aluminosilicates, or combinations thereof.

49. (Withdrawn) The corrosion-inhibiting seal of claim 46 wherein the agent which improves color-fastness is the brightener selected from sulfonic acids, sulfonates, sulfonamides, sulfonic acids, sulfonates, sulfones, cyanides, nonionic surfactants, or combinations thereof.

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50. (Withdrawn) The corrosion-inhibiting seal of claim 44 wherein the color is formed by a dye selected from vat dyes, mordant dyes, lake dyes, disperse dyes, azo dyes, triazine dyes, triphenylmethane dyes, azine dyes, formazan dyes, phthalocyanine dyes, Schiff Base dyes, naturally-occurring dyes, inorganic pigments, or combinations thereof.

51. (Original) The corrosion-inhibiting seal of claim 45 wherein the agent which improves color-fastness is an agent which prevents smudging.

52. (Original) The corrosion-inhibiting seal of claim 51 wherein the agent which prevents smudging is selected from phosphoric acid, metaphosphates, orthophosphates, pyrophosphates, polyphosphates, or combinations thereof.

53. (Original) The corrosion-inhibiting seal of claim 45 wherein the agent which improves color-fastness is a wetting agent.

54. (Original) The corrosion-inhibiting seal of claim 53 further comprising less than about 5 g/L of the wetting agent.

55. (Original) The corrosion-inhibiting seal of claim 53 wherein the wetting agent is a nonionic surfactant.

56. (Withdrawn) A method of making a corrosion-inhibiting sealing bath comprising:  
providing a solvent;  
providing a rare earth source wherein

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the rare earth source is selected from a cerium source, a praseodymium source, a terbium source, or a combination thereof, and

at least one rare earth source is in the tetravalent oxidation state;

providing a valence stabilizer; and

combining the rare earth source and the valence stabilizer to form a rare earth/valence stabilizer complex.

57. (Withdrawn) The method of claim 56 wherein the solvent comprises water.

58. (Withdrawn) The method of claim 56 wherein

the cerium source is selected from a trivalent cerium source, a tetravalent cerium source, or a combination thereof;

the praseodymium source is selected from a trivalent praseodymium source, a tetravalent praseodymium source, or a combination thereof; and

the terbium source is selected from a trivalent terbium source, a tetravalent terbium source, or a combination thereof.

59. (Withdrawn) The method of claim 56 further comprising oxidizing the rare earth source to form the at least one rare earth source in the tetravalent oxidation state.

60. (Withdrawn) The method of claim 59 wherein the rare earth source is oxidized by adding an oxidizer to the corrosion-inhibiting sealing bath.

61. (Withdrawn) The method of claim 60 wherein the oxidizer is a dissolved solid, a liquid, or a gas.

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62. (Withdrawn) The method of claim 60 wherein the oxidizer is selected from peroxides, superoxides, persulfates, perborates, permnitrates, perphosphates, percarbonates, persilicates, peraluminates, pertitanates, perzirconates, permolybdates, pertungstates, pervanadates, organic peroxyacid derivatives, ozone, hypochlorites, chlorates, perchlorates, nitrates, nitrites, vanadates, iodates, hypobromites, chlorites, bromates, permanganates, periodates, dissolved oxygen, dissolved chlorine, dissolved fluorine, or combinations thereof.

63. (Withdrawn) The method of claim 59 wherein the rare earth source is oxidized by electrolysis.

64. (Withdrawn) The method of claim 56 wherein the cerium source is selected from cerium contained within a treated substrate, cerium (III) nitrate, cerium (III) sulfate, cerium (III) perchlorate, cerium (III) chloride, cerium (III) fluoride, cerium (III) bromide, cerium (III) iodide, cerium (III) bromate, cerium (III) fluosilicate, cerium (III) fluotitanate, cerium (III) fluozirconate, cerium (III) fluoborate, cerium (III) fluoaluminate, cerium (III) formate, cerium (III) acetate, cerium (III) propionate, cerium (III) butyrate, cerium (III) glycolate, cerium (III) lactate, cerium (III) sulfonate, cerium (III) alkyl sulfonate, cerium (III) alkoxysulfonate, cerium (III) aromatic sulfonate, cerium (III) aromatoxy sulfonate, cerium (III) sulfamate, cerium (III) alkyl phosphates, cerium (III) acetylacetone, ammonium cerium (III) sulfate, ammonium cerium (III) nitrate, ammonium cerium (III) oxalate, magnesium cerium (III) sulfate, magnesium cerium (III) nitrate, alkali metal cerium (III) nitrate, alkali metal cerium (III) sulfate, cerium (III) carbonate, cerium (III) phosphate, cerium (III) sulfide, cerium (III) fluorocarbonate, cerium (III) benzoate, cerium (III) oxalate, cerium (III) malonate, cerium (III) tartrate, cerium (III) malate, cerium (III) citrate, cerium (III) thiocyanate, cerium (III) salicylate, cerium (III) oxide, cerium (III) hydroxide, cerium (IV) hydroxide species with a hydroxide content of less than or about 50%, cerium (IV) hydroxysulfate, cerium (IV) hydroxychloride, cerium (IV) hydroxynitrate, cerium (IV) hydroxyphosphate, cerium (IV) hydroxyperchlorate, cerium (IV) hydroxyacetate,

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cerium (IV) chloride, cerium (IV) fluoride, cerium (IV) perchlorate, cerium (IV) sulfate, cerium (IV) nitrate, cerium (IV) acetate, cerium (IV) propionate, cerium (IV) butyrate, ammonium cerium (IV) nitrate, ammonium cerium (IV) sulfate, magnesium cerium (IV) nitrate, magnesium cerium (IV) sulfate, alkali metal cerium (IV) nitrate, alkali metal cerium (IV) sulfate, bastnaesite, monazite, xenotime, loparite, lanthanite, rhabdophane, fergusonite, cebaite, aeschynite, lucasite, stillwellite, samarskite, parisite, gadolinite, fluocerite, cerianite, churchite, or combinations thereof.

65. (Withdrawn) The method of claim 56 wherein the praseodymium source is selected from praseodymium contained within a treated substrate, praseodymium nitrate, praseodymium sulfate, praseodymium perchlorate, praseodymium chloride, praseodymium fluoride, praseodymium bromide, praseodymium iodide, praseodymium bromate, praseodymium fluosilicate, praseodymium fluotitanate, praseodymium fluozirconate, praseodymium fluoborate, praseodymium fluoaluminate, praseodymium formate, praseodymium acetate, praseodymium propionate, praseodymium lactate, praseodymium benzenesulfonate, praseodymium acetylacetone, ammonium praseodymium sulfate, ammonium praseodymium nitrate, magnesium praseodymium sulfate, magnesium praseodymium nitrate, alkali metal praseodymium nitrate, alkali metal praseodymium sulfate, bastnaesite, monazite, xenotime, loparite, lanthanite, rhabdophane, fergusonite, cebaite, aeschynite, lucasite, stillwellite, samarskite, parisite, gadolinite, fluocerite, cerianite, churchite, or combinations thereof.

66. (Withdrawn) The method of claim 56 wherein the terbium source is selected from terbium contained within a treated substrate, terbium nitrate, terbium sulfate, terbium perchlorate, terbium chloride, terbium fluoride, terbium bromide, terbium iodide, terbium bromate, terbium fluosilicate, terbium fluotitanate, terbium fluozirconate, terbium fluoborate, terbium fluoaluminate, terbium formate, terbium acetate, terbium propionate, terbium lactate, terbium benzenesulfonate, terbium acetylacetone, ammonium terbium sulfate, ammonium

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terbium nitrate, magnesium terbium sulfate, magnesium terbium nitrate, alkali metal terbium nitrate, alkali metal terbium sulfate, bastnaesite, monazite, xenotime, loparite, lanthanite, rhabdophane, fergusonite, cebaite, aeschynite, lucasite, stillwellite, samarskite, parisite, gadolinite, fluocerite, cerianite, churchite, or combinations thereof.

67. (Withdrawn) The method of claim 56 wherein the valence stabilizer is selected from an inorganic valence stabilizer and an organic valence stabilizer.

68. (Withdrawn) The method of claim 56 further comprising adding a preparative agent to the corrosion-inhibiting sealing bath.

69. (Withdrawn) The method of claim 68 wherein the preparative agent is selected from fluorides, chlorides, bromides, acidic species, hydroxides, or combinations thereof.

70. (Withdrawn) The method of claim 69 wherein the preparative agent is a fluoride.

71. (Withdrawn) The method of claim 70 wherein the fluoride is selected from fluorozirconates, fluorotitanates, fluorosilicates, fluoroaluminates, fluoroborates, fluorogallates, fluoroindates, fluorogermanates, fluorostannates, fluorophosphates, fluoroarsenates, fluoroantimonates, fluorobismuthates, fluorosulfates, fluoroselenates, fluorotellurates, fluoroocuprates, fluoroargentates, fluorozincates, fluorohafnates, fluorovanadates, fluoroniobates, fluorotantalates, fluoromolybdates, fluorotungstates, fluoroyttrates, fluorolanthanates, fluorocerates, fluoromanganates, fluoroferrates, fluoronickelates, fluorocobaltates, potassium fluoride, potassium hydrogen fluoride, sodium fluoride, sodium hydrogen fluoride, lithium fluoride, lithium hydrogen fluoride, ammonium fluoride, ammonium hydrogen fluoride, hydrofluoric acid, dissolved fluorine, organic fluorides, or combinations thereof.

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72. (Withdrawn) The method of claim 69 wherein the preparative agent is an acidic species.

73. (Withdrawn) The method of claim 72 wherein the acidic species is selected from nitric acid, sulfuric acid, phosphoric acid, pyrophosphoric acid, hydrochloric acid, perchloric acid, hydrobromic acid, hydriodic acid, iodic acid, periodic acid, disulfuric acid, selenic acid, telluric acid, polyphosphoric acid, cyclophosphoric acid, phytic acid, boric acid, carboxylic acid, phosphonic acid, sulfonic acid, and acidic metal salts of titanium, zirconium, niobium, tantalum, molybdenum, tungsten, vanadium, aluminum, silicon, tin, antimony, bismuth, tellurium, yttrium, lanthanum, or combinations thereof.

74. (Withdrawn) The method of claim 56 further comprising providing a solubility control agent.

75. (Withdrawn) The method of claim 74 wherein the solubility control agent is selected from a cationic solubility control agent and an anionic solubility control agent.

76. (Withdrawn) The method of claim 75 wherein the solubility control agent is the cationic solubility control agent selected from  $H^+$ ;  $Li^+$ ;  $Na^+$ ;  $K^+$ ;  $Rb^+$ ;  $Cs^+$ ;  $NH_4^+$ ;  $Mg^{+2}$ ;  $Ca^{+2}$ ;  $Sr^{+2}$ ;  $Be^{+2}$ ;  $Ba^{+2}$ ;  $Y^{+3}$ ;  $La^{+3}$ ;  $Ce^{+3}$ ;  $Ce^{+4}$ ;  $Nd^{+3}$ ;  $Pr^{+3}$ ;  $Sc^{+3}$ ;  $Sm^{+3}$ ;  $Eu^{+3}$ ;  $Eu^{+2}$ ;  $Gd^{+3}$ ;  $Tb^{+3}$ ;  $Dy^{+3}$ ;  $Ho^{+3}$ ;  $Er^{+3}$ ;  $Tm^{+3}$ ;  $Yb^{+3}$ ;  $Lu^{+3}$ ;  $Ti^{+4}$ ;  $Zr^{+4}$ ;  $Ti^{+3}$ ;  $Hf^{+4}$ ;  $Nb^{+5}$ ;  $Ta^{+5}$ ;  $Nb^{+4}$ ;  $Ta^{+4}$ ;  $V^{+5}$ ;  $V^{+4}$ ;  $V^{+3}$ ;  $Mo^{+6}$ ;  $W^{+6}$ ;  $Mo^{+5}$ ;  $W^{+5}$ ;  $Mo^{+4}$ ;  $W^{+4}$ ;  $Cr^{+3}$ ;  $Mn^{+2}$ ;  $Mn^{+3}$ ;  $Mn^{+4}$ ;  $Fe^{+2}$ ;  $Fe^{+3}$ ;  $Co^{+2}$ ;  $Co^{+3}$ ;  $Ni^{+2}$ ;  $Ni^{+3}$ ;  $Ni^{+4}$ ;  $Ru^{+2}$ ;  $Ru^{+3}$ ;  $Ru^{+4}$ ;  $Rh^{+3}$ ;  $Ir^{+3}$ ;  $Rh^{+2}$ ;  $Ir^{+2}$ ;  $Pd^{+4}$ ;  $Pt^{+4}$ ;  $Pd^{+2}$ ;  $Pt^{+2}$ ;  $Os^{+4}$ ;  $Cu^+$ ;  $Cu^{+2}$ ;  $Cu^{+3}$ ;  $Ag^+$ ;  $Ag^{+2}$ ;  $Ag^{+3}$ ;  $Au^+$ ;  $Au^{+2}$ ;  $Au^{+3}$ ;  $Zn^{+2}$ ;  $Cd^{+2}$ ;  $Hg^+$ ;  $Hg^{+2}$ ;  $Al^{+3}$ ;  $Ga^{+3}$ ;  $Ga^+$ ;  $In^{+3}$ ;  $In^+$ ;  $Tl^{+3}$ ;  $Tl^+$ ;  $Ge^{+4}$ ;  $Ge^{+2}$ ;  $Sn^{+4}$ ;  $Sn^{+2}$ ;  $Pb^{+4}$ ;  $Pb^{+2}$ ;  $Sb^{+3}$ ;  $Sb^{+5}$ ;  $As^{+3}$ ;  $As^{+5}$ ;  $Bi^{+3}$ ;  $Bi^{+5}$ ; organic compounds containing at least one  $N^+$  site; organic compounds containing at least one phosphonium site; organic compounds containing at least one arsonium site; organic compounds containing at least one stibonium site; organic compounds containing at least one oxonium site; organic compounds containing at least one

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sulfonium site; organic compounds containing at least one selenonium site; organic compounds containing at least one iodonium site; quaternary ammonium compounds having a formula  $NR_4^+$ , where R is an alkyl, aromatic, or acyclic organic constituent; or combinations thereof.

77. (Withdrawn) The method of claim 76 wherein the cationic solubility control agent is selected from  $H^+$ ;  $Li^+$ ;  $Na^+$ ;  $K^+$ ;  $Rb^+$ ;  $Cs^+$ ;  $NH_4^+$ ;  $Mg^{+2}$ ;  $Ca^{+2}$ ;  $Sr^{+2}$ ;  $Y^{+3}$ ;  $La^{+3}$ ;  $Ce^{+3}$ ;  $Ce^{+4}$ ;  $Nd^{+3}$ ;  $Pr^{+3}$ ;  $Sc^{+3}$ ;  $Sm^{+3}$ ;  $Eu^{+3}$ ;  $Eu^{+2}$ ;  $Gd^{+3}$ ;  $Tb^{+3}$ ;  $Dy^{+3}$ ;  $Ho^{+3}$ ;  $Er^{+3}$ ;  $Tm^{+3}$ ;  $Yb^{+3}$ ;  $Lu^{+3}$ ;  $Ti^{+4}$ ;  $Zr^{+4}$ ;  $Ti^{+3}$ ;  $Hf^{+4}$ ;  $Nb^{+5}$ ;  $Ta^{+5}$ ;  $Nb^{+4}$ ;  $Ta^{+4}$ ;  $Mo^{+6}$ ;  $W^{+6}$ ;  $Mo^{+5}$ ;  $W^{+5}$ ;  $Mo^{+4}$ ;  $W^{+4}$ ;  $Mn^{+2}$ ;  $Mn^{+3}$ ;  $Mn^{+4}$ ;  $Fe^{+2}$ ;  $Fe^{+3}$ ;  $Co^{+2}$ ;  $Co^{+3}$ ;  $Ru^{+2}$ ;  $Ru^{+3}$ ;  $Ru^{+4}$ ;  $Rh^{+3}$ ;  $Ir^{+3}$ ;  $Rh^{+2}$ ;  $Ir^{+2}$ ;  $Pd^{+4}$ ;  $Pt^{+4}$ ;  $Pd^{+2}$ ;  $Pt^{+2}$ ;  $Cu^+$ ;  $Cu^{+2}$ ;  $Cu^{+3}$ ;  $Ag^+$ ;  $Ag^{+2}$ ;  $Ag^{+3}$ ;  $Au^+$ ;  $Au^{+2}$ ;  $Au^{+3}$ ;  $Zn^{+2}$ ;  $Al^{+3}$ ;  $Ga^{+3}$ ;  $Ga^+$ ;  $In^{+3}$ ;  $In^+$ ;  $Ge^{+4}$ ;  $Ge^{+2}$ ;  $Sn^{+4}$ ;  $Sn^{+2}$ ;  $Sb^{+3}$ ;  $Sb^{+5}$ ;  $Bi^{+3}$ ;  $Bi^{+5}$ ; organic compounds containing at least one  $N^+$  site; organic compounds containing at least one phosphonium site; organic compounds containing at least one stibonium site; organic compounds containing at least one oxonium site; organic compounds containing at least one sulfonium site; organic compounds containing at least one iodonium site; quaternary ammonium compounds having a formula  $NR_4^+$ , where R is an alkyl, aromatic, or acyclic organic constituent; or combinations thereof.

78. (Withdrawn) The method of claim 75 wherein the solubility control agent is the anionic solubility control agent selected from fluorotitanates, chlorotitanates, fluorozirconates, chlorozirconates, fluoroniobates, chloroniobates, fluorotantalates, chlorotantalates, molybdates, tungstates, permanganates, fluoromanganates, chloromanganates, fluoroferrates, chloroferrates, fluorocobaltates, chlorocobaltates, fluorozincates, chlorozincates, borates, fluoroborates, fluoroaluminates, chloroaluminates, carbonates, silicates, fluorosilicates, fluorostannates, nitrates, nitrites, azides, cyanamides, phosphates, phosphites, phosphonates, phosphinites, thiophosphates, thiophosphites, thiophosphonates, thiophosphinites, fluorophosphates, fluoroantimonates, chloroantimonates, sulfates, sulfites, sulfonates, thiosulfates, dithionites, dithionates, fluorosulfates, tellurates, fluorides, chlorides, chlorates, perchlorates, bromides,

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bromates, iodides, iodates, periodates, heteropolyanions, ferricyanides, ferrocyanides, cyanocobaltates, cyanocuprates, cyanomanganates, cyanates, cyanatoferates, cyanatocobaltates, cyanatocuprates, cyanatomanganates, thiocyanates, thiocyanatoferates, thiocyanatocobaltates, thiocyanatocuprates, thiocyanatomanganates, cyanamides, cyanamidoferates, cyanamidocobaltates, cyanamidocuprates, cyanamidomanganates, nitritoferates, nitritocobaltates, azides, (thio)carboxylates, di(thio)carboxylates, tri(thio)carboxylates, tetra(thio)carboxylates, (thio)phenolates, di(thio)phenolates, tri(thio)phenolates, tetra(thio)phenolates, (thio)phosphonates, di(thio)phosphonates, tri(thio)phosphonates, (thio)phosphonamides, di(thio)phosphonamides, tri(thio)phosphonamides, amino(thio)phosphonates, diamino(thio)phosphonates, triamino(thio)phosphonates, imino(thio)phosphonates, diimino(thio)phosphonates, (thio)sulfonates, di(thio)sulfonates, tri(thio)sulfonates, (thio)sulfonamides, di(thio)sulfonamides, tri(thio)sulfonamides, amino(thio)sulfonates, diamino(thio)sulfonates, triamino(thio)sulfonates, imino(thio)sulfonates, diimino(thio)sulfonates, (thio)borates, di(thio)borates, (thio)boronates, organic silicates, stibonates, cyanides, cyanochromates, cyanonickelates, cyanatochromates, cyanatonickelates, thiocyanatochromates, thiocyanatonickelates, cyanamidochromates, cyanamidonickelates, nitritonickelates, arsonates, diarsonates, triarsonates, organic selenates, diselenates, triselenates, arsenates, arsenites, fluoroarsenates, chloroarsenates, selenates, selenites, fluorothallates, chlorothallates, iodomercury anions, chloromercurates, bromomercurates, osmates, fluoronickelates, chromates, Reinecke's salt, vanadates, or combinations thereof.

79. (Withdrawn) The method of claim 78 wherein the anionic solubility control agent is selected from fluorotitanates, chlorotitanates, fluorozirconates, chlorozirconates, fluoroniobates, chloroniobates, fluorotantalates, chlorotantalates, molybdates, tungstates, permanganates, fluoromanganates, chloromanganates, fluoroferates, chloroferrates, fluorocobaltates, chlorocobaltates, fluorozincates, chlorozincates, borates, fluoroborates, fluoroaluminates, chloroaluminates, carbonates, silicates, fluorosilicates, fluorostannates, nitrates, nitrites, azides,

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cyanamides, phosphates, phosphites, phosphonates, phosphinites, thiophosphates, thiophosphites, thiophosphonates, thiophosphinites, fluorophosphates, fluoroantimonates, chloroantimonates, sulfates, sulfites, sulfonates, thiosulfates, dithionites, dithionates, fluorosulfates, tellurates, fluorides, chlorides, chlorates, perchlorates, bromides, bromates, iodides, iodates, periodates, heteropolyanions, ferricyanides, ferrocyanides, cyanocobaltates, cyanocuprates, cyanomanganates, cyanates, cyanatoferrates, cyanatocobaltates, cyanatocuprates, cyanatomanganates, thiocyanates, thiocyanatoferrates, thiocyanatocobaltates, thiocyanatocuprates, thiocyanatomanganates, cyanamides, cyanamidoferates, cyanamidocobaltates, cyanamidocuprates, cyanamidomanganates, nitritoferates, nitritocobaltates, azides, (thio)carboxylates, di(thio)carboxylates, tri(thio)carboxylates, tetra(thio)carboxylates, (thio)phenolates, di(thio)phenolates, tri(thio)phenolates, tetra(thio)phenolates, (thio)phosphonates, di(thio)phosphonates, tri(thio)phosphonates, (thio)phosphonamides, di(thio)phosphonamides, tri(thio)phosphonamides, amino(thio)phosphonates, diamino(thio)phosphonates, triamino(thio)phosphonates, imino(thio)phosphonates, diimino(thio)phosphonates, (thio)sulfonates, di(thio)sulfonates, tri(thio)sulfonates, (thio)sulfonamides, di(thio)sulfonamides, tri(thio)sulfonamides, amino(thio)sulfonates, diamino(thio)sulfonates, triamino(thio)sulfonates, imino(thio)sulfonates, diimino(thio)sulfonates, (thio)borates, di(thio)borates, (thio)boronates, organic silicates, stibonates, or combinations thereof.

80. (Withdrawn) The method of claim 56 wherein the corrosion-inhibiting sealing bath has a concentration of cerium, praseodymium, or terbium of between a minimum concentration of about  $1 \times 10^{-4}$  moles/liter cerium, praseodymium, or terbium and a maximum concentration which is a maximum solubility of the cerium, praseodymium, or terbium source in the solvent at a temperature of the sealing bath.

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81. (Withdrawn) The method of claim 60 wherein the corrosion-inhibiting sealing bath has a concentration of oxidizer of between a minimum concentration wherein a majority of the cerium, praseodymium, or terbium is oxidized to a tetravalent oxidation state and a maximum concentration which is the maximum solubility of the oxidizer in the solvent at a temperature of the sealing bath.

82. (Withdrawn) The method of claim 70 wherein the corrosion-inhibiting sealing bath has a concentration of preparative agent between a minimum concentration, wherein there is a fluoride-to-cerium, fluoride-to-praseodymium, or fluoride-to-terbium ratio of 0.05 and a maximum concentration, which is a maximum solubility of the preparative agent in the solvent at a temperature of the sealing bath.

83. (Withdrawn) The method of claim 72 wherein the corrosion-inhibiting sealing bath has a concentration of preparative agent between a minimum concentration, wherein there is an acidic species-to-cerium, acidic species-to-praseodymium, or acidic species-to-terbium ratio of 0.05 and a maximum concentration, which is a maximum solubility of the preparative agent in the solvent at a temperature of the sealing bath.

84. (Withdrawn) The method of claim 68 wherein the corrosion-inhibiting sealing bath has a concentration of the preparative agent of less than 5.0 M.

85. (Withdrawn) The method of claim 84 wherein the corrosion-inhibiting sealing bath has a concentration of the preparative agent of less than 0.5 M.

86. (Withdrawn) The method of claim 56 wherein the corrosion-inhibiting sealing bath has a concentration of valence stabilizer between a minimum concentration wherein there is a valence stabilizer-to-cerium, valence stabilizer-to-praseodymium, or valence stabilizer-to-terbium ratio

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of 0.010 and a maximum concentration which is a maximum solubility of the valence stabilizer in the solvent at a temperature of the sealing bath.

87. (Withdrawn) The method of claim 56 wherein the corrosion-inhibiting sealing bath has a pH of between about 0 and about 9.

88. (Withdrawn) The method of claim 87 wherein the corrosion-inhibiting sealing bath has a pH of between about 1 and about 8.

89. (Withdrawn) The method of claim 56 wherein the corrosion-inhibiting sealing bath has a temperature of between about 5°C and about 100°C.

90. (Withdrawn) The method of claim 89 wherein the corrosion-inhibiting sealing bath has a temperature of between about 20°C and about 50°C.

91. (Withdrawn) A method of applying a corrosion-inhibiting seal to a barrier film, comprising:

providing the barrier film to be treated;

contacting the barrier film with a first corrosion-inhibiting sealing bath comprising a first solvent and a rare earth source wherein

the rare earth source is selected from a cerium source, a praseodymium source, a terbium source, or a combination thereof, and

at least one rare earth source is in the tetravalent oxidation state; and

contacting the barrier film with a valence stabilizer to form a seal comprising a rare earth/valence stabilizer complex.

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92. (Withdrawn) The method of claim 91 wherein the valence stabilizer is added to the first corrosion-inhibiting sealing bath.

93. (Withdrawn) The method of claim 91 wherein the first solvent comprises water.

94. (Withdrawn) The method of claim 91 further comprising providing a second corrosion-inhibiting sealing bath comprising a second solvent and the valence stabilizer.

95. (Withdrawn) The method of claim 94 wherein the second solvent comprises water.

96. (Withdrawn) The method of claim 91 wherein the barrier film is a coating selected from oxides, hydroxides, phosphates, carbonates, oxalates, silicates, aluminates, borates, polymers, or combinations thereof.

97. (Withdrawn) The method of claim 91 wherein  
the cerium source is selected from a trivalent cerium source, a tetravalent cerium source, or a combination thereof,  
the praseodymium source is selected from a trivalent praseodymium source, a tetravalent praseodymium source, or a combination thereof, and  
the terbium source is selected from a trivalent terbium source, a tetravalent terbium source, or a combination thereof.

98. (Withdrawn) The method of claim 91 further comprising oxidizing the rare earth source to form the at least one rare earth source in the tetravalent oxidation state.

99. (Withdrawn) The method of claim 98 wherein the rare earth source is oxidized in the first corrosion-inhibiting sealing bath.

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100. (Withdrawn) The method of claim 98 wherein the rare earth source is oxidized in the seal.

101. (Withdrawn) The method of claim 98 wherein the rare earth source is oxidized by adding an oxidizer to the first corrosion-inhibiting sealing bath.

102. (Withdrawn) The method of claim 98 wherein the rare earth source is oxidized by electrolysis.

103. (Withdrawn) The method of claim 101 wherein the oxidizer is selected from a dissolved solid, a liquid, and a gas.

104. (Withdrawn) The method of claim 101 wherein the oxidizer is selected from peroxides, superoxides, persulfates, perborates, pernitrates, perphosphates, percarbonates, persilicates, peraluminates, pertitanates, perzirconates, permolybdates, pertungstates, pervanadates, organic peroxyacid derivatives, ozone, hypochlorites, chlorates, perchlorates, nitrates, nitrites, vanadates, iodates, hypobromites, chlorites, bromates, permanganates, periodates, dissolved oxygen, dissolved chlorine, dissolved fluorine, or combinations thereof.

105. (Withdrawn) The method of claim 91 wherein the cerium source is selected from cerium contained within a treated substrate, cerium (III) nitrate, cerium (III) sulfate, cerium (III) perchlorate, cerium (III) chloride, cerium (III) fluoride, cerium (III) bromide, cerium (III) iodide, cerium (III) bromate, cerium (III) fluosilicate, cerium (III) fluotitanate, cerium (III) fluozirconate, cerium (III) fluoborate, cerium (III) fluoaluminate, cerium (III) formate, cerium (III) acetate, cerium (III) propionate, cerium (III) butyrate, cerium (III) glycolate, cerium (III) lactate, cerium (III) sulfonate, cerium (III) alkyl sulfonate, cerium (III) alkoxysulfonate, cerium

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(III) aromatic sulfonate, cerium (III) aromatoxy sulfonate, cerium (III) sulfamate, cerium (III) alkyl phosphates, cerium (III) acetylacetone, ammonium cerium (III) sulfate, ammonium cerium (III) nitrate, ammonium cerium (III) oxalate, magnesium cerium (III) sulfate, magnesium cerium (III) nitrate, alkali metal cerium (III) nitrate, alkali metal cerium (III) sulfate, cerium (III) carbonate, cerium (III) phosphate, cerium (III) sulfide, cerium (III) fluorocarbonate, cerium (III) benzoate, cerium (III) oxalate, cerium (III) malonate, cerium (III) tartrate, cerium (III) malate, cerium (III) citrate, cerium (III) thiocyanate, cerium (III) salicylate, cerium (III) oxide, cerium (III) hydroxide, cerium (IV) hydroxide species with a hydroxide content of less than or about 50%, cerium (IV) hydroxysulfate, cerium (IV) hydroxychloride, cerium (IV) hydroxynitrate, cerium (IV) hydroxyphosphate, cerium (IV) hydroxypchlorate, cerium (IV) hydroxyacetate, cerium (IV) chloride, cerium (IV) fluoride, cerium (IV) perchlorate, cerium (IV) sulfate, cerium (IV) nitrate, cerium (IV) acetate, cerium (IV) propionate, cerium (IV) butyrate, ammonium cerium (IV) nitrate, ammonium cerium (IV) sulfate, magnesium cerium (IV) nitrate, magnesium cerium (IV) sulfate, alkali metal cerium (IV) nitrate, alkali metal cerium (IV) sulfate, bastnaesite, monazite, xenotime, loparite, lanthanite, rhabdophane, fergusonite, cebaite, aeschynite, lucasite, stillwellite, samarskite, parisite, gadolinite, fluocerite, cerianite, churchite, or combinations thereof.

106. (Withdrawn) The method of claim 91 wherein the praseodymium source is selected from praseodymium contained within a treated substrate, praseodymium nitrate, praseodymium sulfate, praseodymium perchlorate, praseodymium chloride, praseodymium fluoride, praseodymium bromide, praseodymium iodide, praseodymium bromate, praseodymium fluosilicate, praseodymium fluotitanate, praseodymium fluozirconate, praseodymium fluoborate, praseodymium fluoaluminate, praseodymium formate, praseodymium acetate, praseodymium propionate, praseodymium lactate, praseodymium benzenesulfonate, praseodymium acetylacetone, ammonium praseodymium sulfate, ammonium praseodymium nitrate, magnesium praseodymium sulfate, magnesium praseodymium nitrate, alkali metal

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praseodymium nitrate, alkali metal praseodymium sulfate, bastnaesite, monazite, xenotime, loparite, lanthanite, rhabdophane, fergusonite, cebaite, aeschynite, lucasite, stillwellite, samarskite, parisite, gadolinite, fluocerite, cerianite, churchite, or combinations thereof.

107. (Withdrawn) The method of claim 91 wherein the terbium source is selected from terbium contained within a treated substrate, terbium nitrate, terbium sulfate, terbium perchlorate, terbium chloride, terbium fluoride, terbium bromide, terbium iodide, terbium bromate, terbium fluosilicate, terbium fluotitanate, terbium fluozirconate, terbium fluoborate, terbium fluoaluminate, terbium formate, terbium acetate, terbium propionate, terbium lactate, terbium benzenesulfonate, terbium acetylacetone, ammonium terbium sulfate, ammonium terbium nitrate, magnesium terbium sulfate, magnesium terbium nitrate, alkali metal terbium nitrate, alkali metal terbium sulfate, bastnaesite, monazite, xenotime, loparite, lanthanite, rhabdophane, fergusonite, cebaite, aeschynite, lucasite, stillwellite, samarskite, parisite, gadolinite, fluocerite, cerianite, churchite, or combinations thereof.

108. (Withdrawn) The method of claim 91 wherein the valence stabilizer is selected from an inorganic valence stabilizer and an organic valence stabilizer.

109. (Withdrawn) The method of claim 91 wherein the barrier film is contacted with the first corrosion-inhibiting sealing bath by a process selected from immersion, spraying, fogging, wiping, and dipping.

110. (Withdrawn) The method of claim 94 wherein the barrier film is contacted with the second corrosion-inhibiting sealing bath by a process selected from immersion, spraying, fogging, wiping, and dipping.

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111. (Withdrawn) The method of claim 91 further comprising adding a preparative agent to the first corrosion-inhibiting sealing bath.

112. (Withdrawn) The method of claim 111 wherein the preparative agent is selected from fluorides, chlorides, bromides, acidic species, hydroxides, or combinations thereof.

113. (Withdrawn) The method of claim 112 wherein the preparative agent is a fluoride.

114. (Withdrawn) The method of claim 113 wherein the fluoride is selected from fluorozirconates, fluorotitanates, fluorosilicates, fluoroaluminates, fluoroborates, fluorogallates, fluoroindates, fluorogermanates, fluorostannates, fluorophosphates, fluoroarsenates, fluoroantimonates, fluorobismuthates, fluorosulfates, fluoroselenates, fluorotellurates, fluorocuprates, fluoroargentates, fluorozincates, fluorohafnates, fluorovanadates, fluoroniobates, fluorotantalates, fluoromolybdates, fluorotungstates, fluoroyttrates, fluorolanthanates, fluorocerates, fluoromanganates, fluoroferrates, fluoronickelates, fluorocobaltates, potassium fluoride, potassium hydrogen fluoride, sodium fluoride, sodium hydrogen fluoride, lithium fluoride, lithium hydrogen fluoride, ammonium fluoride, ammonium hydrogen fluoride, hydrofluoric acid, dissolved fluorine, organic fluorides, or combinations thereof.

115. (Withdrawn) The method of claim 112 wherein the preparative agent is an acidic species.

116. (Withdrawn) The method of claim 115 wherein the acidic species is selected from nitric acid, sulfuric acid, phosphoric acid, pyrophosphoric acid, hydrochloric acid, perchloric acid, hydrobromic acid, hydriodic acid, iodic acid, periodic acid, disulfuric acid, selenic acid, telluric acid, polyphosphoric acid, cyclophosphoric acid, phytic acid, boric acid, carboxylic acid, phosphonic acid, sulfonic acid, and acidic metal salts of titanium, zirconium, niobium, tantalum,

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molybdenum, tungsten, vanadium, aluminum, silicon, tin, antimony, bismuth, tellurium, yttrium, lanthanum, or combinations thereof.

117. (Withdrawn) The method of claim 91 further comprising contacting the seal comprising the rare earth/valence stabilizer complex with a solubility control agent.

118. (Withdrawn) The method of claim 117 wherein the solubility control agent is added to the first corrosion-inhibiting sealing bath.

119. (Withdrawn) The method of claim 117 further comprising a second corrosion-inhibiting sealing bath comprising a second solvent, the valence stabilizer, and the solubility control agent.

120. (Withdrawn) The method of claim 119 wherein the second solvent is water.

121. (Withdrawn) The method of claim 117 wherein the solubility control agent is selected from a cationic solubility control agent and an anionic solubility control agent.

122. (Withdrawn) The method of claim 121 wherein the solubility control agent is a cationic solubility control agent selected from  $H^+$ ;  $Li^+$ ;  $Na^+$ ;  $K^+$ ;  $Rb^+$ ;  $Cs^+$ ;  $NH_4^+$ ;  $Mg^{+2}$ ;  $Ca^{+2}$ ;  $Sr^{+2}$ ;  $Be^{+2}$ ;  $Ba^{+2}$ ;  $Y^{+3}$ ;  $La^{+3}$ ;  $Ce^{+3}$ ;  $Ce^{+4}$ ;  $Nd^{+3}$ ;  $Pr^{+3}$ ;  $Sc^{+3}$ ;  $Sm^{+3}$ ;  $Eu^{+3}$ ;  $Eu^{+2}$ ;  $Gd^{+3}$ ;  $Tb^{+3}$ ;  $Dy^{+3}$ ;  $Ho^{+3}$ ;  $Er^{+3}$ ;  $Tm^{+3}$ ;  $Yb^{+3}$ ;  $Lu^{+3}$ ;  $Ti^{+4}$ ;  $Zr^{+4}$ ;  $Ti^{+3}$ ;  $Hf^{+4}$ ;  $Nb^{+5}$ ;  $Ta^{+5}$ ;  $Nb^{+4}$ ;  $Ta^{+4}$ ;  $V^{+5}$ ;  $V^{+4}$ ;  $V^{+3}$ ;  $Mo^{+6}$ ;  $W^{+6}$ ;  $Mo^{+5}$ ;  $W^{+5}$ ;  $Mo^{+4}$ ;  $W^{+4}$ ;  $Cr^{+3}$ ;  $Mn^{+2}$ ;  $Mn^{+3}$ ;  $Mn^{+4}$ ;  $Fe^{+2}$ ;  $Fe^{+3}$ ;  $Co^{+2}$ ;  $Co^{+3}$ ;  $Ni^{+2}$ ;  $Ni^{+3}$ ;  $Ni^{+4}$ ;  $Ru^{+2}$ ;  $Ru^{+3}$ ;  $Ru^{+4}$ ;  $Rh^{+3}$ ;  $Ir^{+3}$ ;  $Rh^{+2}$ ;  $Ir^{+2}$ ;  $Pd^{+4}$ ;  $Pt^{+4}$ ;  $Pd^{+2}$ ;  $Pt^{+2}$ ;  $Os^{+4}$ ;  $Cu^+$ ;  $Cu^{+2}$ ;  $Cu^{+3}$ ;  $Ag^+$ ;  $Ag^{+2}$ ;  $Ag^{+3}$ ;  $Au^+$ ;  $Au^{+2}$ ;  $Au^{+3}$ ;  $Zn^{+2}$ ;  $Cd^{+2}$ ;  $Hg^+$ ;  $Hg^{+2}$ ;  $Al^{+3}$ ;  $Ga^{+3}$ ;  $Ga^+$ ;  $In^{+3}$ ;  $In^+$ ;  $Tl^{+3}$ ;  $Tl^+$ ;  $Ge^{+4}$ ;  $Ge^{+2}$ ;  $Sn^{+4}$ ;  $Sn^{+2}$ ;  $Pb^{+4}$ ;  $Pb^{+2}$ ;  $Sb^{+3}$ ;  $Sb^{+5}$ ;  $As^{+3}$ ;  $As^{+5}$ ;  $Bi^{+3}$ ;  $Bi^{+5}$ ; organic compounds containing at least one  $N^+$  site; organic compounds containing at least one phosphonium site; organic compounds containing at least one arsonium site; organic compounds containing at least one stibonium site; organic

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compounds containing at least one oxonium site; organic compounds containing at least one sulfonium site; organic compounds containing at least one selenonium site; organic compounds containing at least one iodonium site; quarternary ammonium compounds having a formula of  $NR_4^+$ , where R is an alkyl, aromatic, or acyclic organic constituent; or combinations thereof.

123. (Withdrawn) The method of claim 122 wherein the cationic solubility control agent is selected from  $H^+$ ;  $Li^+$ ;  $Na^+$ ;  $K^+$ ;  $Rb^+$ ;  $Cs^+$ ;  $NH_4^+$ ;  $Mg^{+2}$ ;  $Ca^{+2}$ ;  $Sr^{+2}$ ;  $Y^{+3}$ ;  $La^{+3}$ ;  $Ce^{+3}$ ;  $Ce^{+4}$ ;  $Nd^{+3}$ ;  $Pr^{+3}$ ;  $Sc^{+3}$ ;  $Sm^{+3}$ ;  $Eu^{+3}$ ;  $Eu^{+2}$ ;  $Gd^{+3}$ ;  $Tb^{+3}$ ;  $Dy^{+3}$ ;  $Ho^{+3}$ ;  $Er^{+3}$ ;  $Tm^{+3}$ ;  $Yb^{+3}$ ;  $Lu^{+3}$ ;  $Ti^{+4}$ ;  $Zr^{+4}$ ;  $Ti^{+3}$ ;  $Hf^{+4}$ ;  $Nb^{+5}$ ;  $Ta^{+5}$ ;  $Nb^{+4}$ ;  $Ta^{+4}$ ;  $Mo^{+6}$ ;  $W^{+6}$ ;  $Mo^{+5}$ ;  $W^{+5}$ ;  $Mo^{+4}$ ;  $W^{+4}$ ;  $Mn^{+2}$ ;  $Mn^{+3}$ ;  $Mn^{+4}$ ;  $Fe^{+2}$ ;  $Fe^{+3}$ ;  $Co^{+2}$ ;  $Co^{+3}$ ;  $Ru^{+2}$ ;  $Ru^{+3}$ ;  $Ru^{+4}$ ;  $Rh^{+3}$ ;  $Ir^{+3}$ ;  $Rh^{+2}$ ;  $Ir^{+2}$ ;  $Pd^{+4}$ ;  $Pt^{+4}$ ;  $Pd^{+2}$ ;  $Pt^{+2}$ ;  $Cu^+$ ;  $Cu^{+2}$ ;  $Cu^{+3}$ ;  $Ag^+$ ;  $Ag^{+2}$ ;  $Ag^{+3}$ ;  $Au^+$ ;  $Au^{+2}$ ;  $Au^{+3}$ ;  $Zn^{+2}$ ;  $Cd^{+2}$ ;  $Al^{+3}$ ;  $Ga^{+3}$ ;  $Ga^+$ ;  $In^{+3}$ ;  $In^+$ ;  $Ge^{+4}$ ;  $Ge^{+2}$ ;  $Sn^{+4}$ ;  $Sn^{+2}$ ;  $Sb^{+3}$ ;  $Sb^{+5}$ ;  $Bi^{+3}$ ;  $Bi^{+5}$ ; organic compounds containing at least one  $N^+$  site; organic compounds containing at least one phosphonium site; organic compounds containing at least one stibonium site; organic compounds containing at least one oxonium site; organic compounds containing at least one sulfonium site; organic compounds containing at least one iodonium site; quarternary ammonium compounds having a formula of  $NR_4^+$ , where R is an alkyl, aromatic, or acyclic organic constituent; or combinations thereof.

124. (Withdrawn) The method of claim 121 wherein the solubility control agent is the anionic solubility control agent selected from fluorotitanates, chlorotitanates, fluorozirconates, chlorozirconates, fluoroniobates, chloroniobates, fluorotantalates, chlorotantalates, molybdates, tungstates, permanganates, fluoromanganates, chloromanganates, fluoroferrates, chloroferrates, fluorocobaltates, chlorocobaltates, fluorozincates, chlorozincates, borates, fluoroborates, fluoroaluminates, chloroaluminates, carbonates, silicates, fluorosilicates, fluorostannates, nitrates, nitrites, azides, cyanamides, phosphates, phosphites, phosphonates, phosphinites, thiophosphates, thiophosphites, thiophosphonates, thiophosphinites, fluorophosphates, fluoroantimonates, chloroantimonates, sulfates, sulfites, sulfonates, thiosulfates, dithionites,

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dithionates, fluorosulfates, tellurates, fluorides, chlorides, chlorates, perchlorates, bromides, bromates, iodides, iodates, periodates, heteropolyanions, ferricyanides, ferrocyanides, cyanocobaltates, cyanocuprates, cyanomanganates, cyanates, cyanatoferrates, cyanatocobaltates, cyanatocuprates, cyanatomanganates, thiocyanates, thiocyanatoferrates, thiocyanatocobaltates, thiocyanatocuprates, thiocyanatomanganates, cyanamides, cyanamidoferrates, cyanamidocobaltates, cyanamidocuprates, cyanamidomanganates, nitritoferrates, nitritocobaltates, azides, (thio)carboxylates, di(thio)carboxylates, tri(thio)carboxylates, tetra(thio)carboxylates, (thio)phenolates, di(thio)phenolates, tri(thio)phenolates, tetra(thio)phenolates, (thio)phosphonates, di(thio)phosphonates, tri(thio)phosphonates, (thio)phosphonamides, di(thio)phosphonamides, tri(thio)phosphonamides, amino(thio)phosphonates, diamino(thio)phosphonates, triamino(thio)phosphonates, imino(thio)phosphonates, diimino(thio)phosphonates, (thio)sulfonates, di(thio)sulfonates, tri(thio)sulfonates, (thio)sulfonamides, di(thio)sulfonamides, tri(thio)sulfonamides, amino(thio)sulfonates, diamino(thio)sulfonates, triamino(thio)sulfonates, imino(thio)sulfonates, diimino(thio)sulfonates, (thio)borates, di(thio)borates, (thio)boronates, organic silicates, stibonates, cyanides, cyanochromates, cyanonickelates, cyanatochromates, cyanatonickelates, thiocyanatochromates, thiocyanatonickelates, cyanamidochromates, cyanamidonickelates, nitritonickelates, arsonates, diarsonates, triarsonates, organic selenates, diselenates, triselenates, arsenates, arsenites, fluoroarsenates, chloroarsenates, selenates, selenites, fluorothallates, chlorothallates, iodomercury anions, chloromercurates, bromomercurates, osmates, fluoronickelates, chromates, Reinecke's salt, vanadates, or combinations thereof.

125. (Withdrawn) The method of claim 124 wherein the anionic solubility control agent is selected from fluorotitanates, chlorotitanates, fluorozirconates, chlorozirconates, fluoroniobates, chloroniobates, fluorotantalates, chlorotantalates, molybdates, tungstates, permanganates, fluoromanganates, chloromanganates, fluoroferates, chloroferrates, fluorocobaltates, chlorocobaltates, fluorozincates, chlorozincates, borates, fluoroborates, fluoroaluminates,

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chloroaluminates, carbonates, silicates, fluorosilicates, fluorostannates, nitrates, nitrites, azides, cyanamides, phosphates, phosphites, phosphonates, phosphinites, thiophosphates, thiophosphites, thiophosphonates, thiophosphinites, fluorophosphates, fluoroantimonates, chloroantimonates, sulfates, sulfites, sulfonates, thiosulfates, dithionites, dithionates, fluorosulfates, tellurates, fluorides, chlorides, chlorates, perchlorates, bromides, bromates, iodides, iodates, periodates, heteropolyanions, ferricyanides, ferrocyanides, cyanocobaltates, cyanocuprates, cyanomanganates, cyanates, cyanoferrates, cyanatocobaltates, cyanatocuprates, cyanatomanganates, thiocyanates, thiocyanatoferates, thiocyanatocobaltates, thiocyanatocuprates, thiocyanatomanganates, cyanamides, cyanamidoferates, cyanamidocobaltates, cyanamidocuprates, cyanamidoferates, cyanamidocobaltates, cyanamidocuprates, cyanamidomanganates, nitroferrates, nitritocobaltates, azides, (thio)carboxylates, di(thio)carboxylates, tri(thio)carboxylates, tetra(thio)carboxylates, (thio)phenolates, di(thio)phenolates, tri(thio)phenolates, tetra(thio)phenolates, (thio)phosphonates, di(thio)phosphonates, tri(thio)phosphonates, (thio)phosphonamides, di(thio)phosphonamides, tri(thio)phosphonamides, amino(thio)phosphonates, diamino(thio)phosphonates, triamino(thio)phosphonates, imino(thio)phosphonates, diimino(thio)phosphonates, (thio)sulfonates, di(thio)sulfonates, tri(thio)sulfonates, (thio)sulfonamides, di(thio)sulfonamides, tri(thio)sulfonamides, amino(thio)sulfonates, diamino(thio)sulfonates, triamino(thio)sulfonates, imino(thio)sulfonates, diimino(thio)sulfonates, (thio)borates, di(thio)borates, (thio)boronates, organic silicates, stibonates, or combinations thereof.

126. (Withdrawn) The method of claim 91 wherein the first corrosion-inhibiting sealing bath has a concentration of cerium, praseodymium, or terbium of between a minimum concentration of about  $1 \times 10^{-4}$  moles/liter cerium, praseodymium, or terbium and a maximum concentration which is a maximum solubility of the cerium, praseodymium, or terbium source in the first solvent at a temperature of the sealing bath.

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127. (Withdrawn) The method of claim 98 wherein the first corrosion-inhibiting sealing bath has a concentration of oxidizer between a minimum concentration wherein a majority of the cerium, praseodymium, or terbium is oxidized to a tetravalent oxidation state and a maximum concentration which is the maximum solubility of the oxidizer in the first solvent at a temperature of the sealing bath.

128. (Withdrawn) The method of claim 113 wherein the first corrosion-inhibiting sealing bath has a concentration of preparative agent between a minimum concentration, wherein there is a fluoride-to-cerium, fluoride-to-praseodymium, or fluoride-to-terbium ratio of 0.05 and a maximum concentration, which is a maximum solubility of the preparative agent in the first solvent at a temperature of the sealing bath.

129. (Withdrawn) The method of claim 115 wherein the first corrosion-inhibiting sealing bath has a concentration of preparative agent between a minimum concentration, wherein there is an acidic species-to-cerium, acidic species-to-praseodymium, or acidic species-to-terbium ratio of 0.05 and a maximum concentration, which is a maximum solubility of the preparative agent in the first solvent at a temperature of the sealing bath.

130. (Withdrawn) The method of claim 111 wherein the first corrosion-inhibiting sealing bath has a concentration of preparative agent of less than 5.0 M.

131. (Withdrawn) The method of claim 130 wherein the first corrosion-inhibiting sealing bath has a concentration of preparative agent of less than 0.5 M.

132. (Withdrawn) The method of claim 91 wherein the first corrosion-inhibiting sealing bath has a concentration of valence stabilizer between a minimum concentration wherein there is a valence stabilizer-to-cerium, valence stabilizer-to-praseodymium, or valence stabilizer-to-

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terbium ratio of 0.010 and a maximum concentration which is a maximum solubility of the valence stabilizer in the first solvent at a temperature of the sealing bath.

133. (Withdrawn) The method of claim 91 wherein the first corrosion-inhibiting sealing bath has a pH of between about 0 and about 9.

134. (Withdrawn) The method of claim 133 wherein the first corrosion-inhibiting sealing bath has a pH of between about 1 and about 8.

135. (Withdrawn) The method of claim 91 wherein the first corrosion-inhibiting sealing bath has a temperature of between about 5°C and about 100°C.

136. (Withdrawn) The method of claim 135 wherein the first corrosion-inhibiting sealing bath has a temperature of between about 20°C and about 50°C.

137. (Currently Amended) A solid corrosion-inhibiting seal for a barrier film formed on a coating selected from anodic coatings, phosphating coatings, or black oxide coatings, the solid corrosion-inhibiting seal comprising a rare earth element and a valence stabilizer combined to form a rare earth/valence stabilizer complex within the solid corrosion-inhibiting seal, wherein the rare earth element is selected from cerium, praseodymium, terbium, or combinations thereof, at least one rare earth element is in the tetravalent oxidation state, and the rare earth/valence stabilizer complex is sparingly soluble in water at about 25°C and about 760 Torr.

138. (Withdrawn) A method of making a corrosion-inhibiting sealing bath comprising:  
providing a solvent;  
providing a rare earth source wherein

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the rare earth source is selected from a cerium source, a praseodymium source, a terbium source, or a combination thereof, and  
at least one rare earth source is in the tetravalent oxidation state;  
providing a valence stabilizer; and  
combining the rare earth source and the valence stabilizer to form a rare earth/valence stabilizer complex, wherein the rare earth/valence stabilizer complex is sparingly soluble in water at about 25°C and about 760 Torr.

139. (Withdrawn) A method of applying a corrosion-inhibiting seal to a barrier film comprising:

providing the barrier film to be treated;  
contacting the barrier film with a first corrosion-inhibiting sealing bath comprising a first solvent and a rare earth source wherein  
the rare earth source is selected from a cerium source, a praseodymium source, a terbium source, or a combination thereof, and  
at least one rare earth source is in the tetravalent oxidation state; and  
contacting the barrier film with a valence stabilizer to form a seal comprising a rare earth/valence stabilizer complex, wherein the rare earth/valence stabilizer complex is sparingly soluble in water at about 25°C and about 760 Torr.